Oxford A Level Sciences

AQA Physics

1 Which type of error occurs due to unpredictable variation around the true value?

Circle one answer.

A measurement error	В	random error	
C systematic error	D	zero error	[1]

2 Which type of error occurs due to using an ammeter that reads 0.03 A when there is no current passing through it?

Circle one answer.

- A measurement errorBrandom errorC systematic errorDzero error[1]
- 3 What is the specific charge of a proton?

Circle **one** answer.

- **A** $1.76 \times 10^{11} \,\mathrm{C \, kg^{-1}}$
- **B** $1.67 \times 10^{-11} \,\mathrm{C \, kg^{-1}}$
- **C** $9.58 \times 10^7 \,\mathrm{C \, kg^{-1}}$
- **D** $9.11 \times 10^{-31} \,\mathrm{C \, kg^{-1}}$ [1]
- 4 Circle one answer. An alpha particle is made up of:

A 2 electrons & 2 protons	В	2 electrons & 2 neutrons	
C 2 neutrons & 2 protons	D	2 neutrons, 2 protons, & 2 electrons	[1]

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5 Match these measurement terms to their definitions.



5 Draw a line from each quantity to its SI unit.

mass	degree Celsius
	second
length	centimetre
5	ampere
time	kilogram
	minute
current	volt
	kelvin
temperature	metre
	gram

6 Complete the table below

prefix	exponent
micro	
	10 ⁻³
centi	
	10 ⁻⁹
kilo	

[3]

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[2]

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[2]

7 Write the types of electromagnetic radiation in order of decreasing wavelength.

- 8 Complete the sentence below.
 - Carbon-12 and carbon-14 are ______ of carbon. The difference between them is
 - the number of _____. They have the same number of protons and _____. [2]
- **9** Carbon-14 decays into nitrogen-14 through beta decay. Name the particle released and write its symbol.

particle released =	symbol =	[2]
	Symbol =	[~]

- **10 a** Write the equation for the wavelength of electromagnetic radiation.
 - **b** Calculate the frequency of light with a wavelength of 725 nm

frequency =	_ units	
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[1]

[3]

11 Resolve these vectors into their horizontal and vertical components using trigonometry.



horizontal component =		Ν
vertical component =	Ν	

[2]



horizontal component = _____ N vertical component = _____ N

[2]

12 Complete the table below.

	Mass (kg)	Absolute Charge (C)	Relative Charge
proton	1.67 x 10 ⁻²⁷		1
neutron		0	
electron			

13 Name the four fundamental interactions.

[2]

[3]

Total = 30 marks